

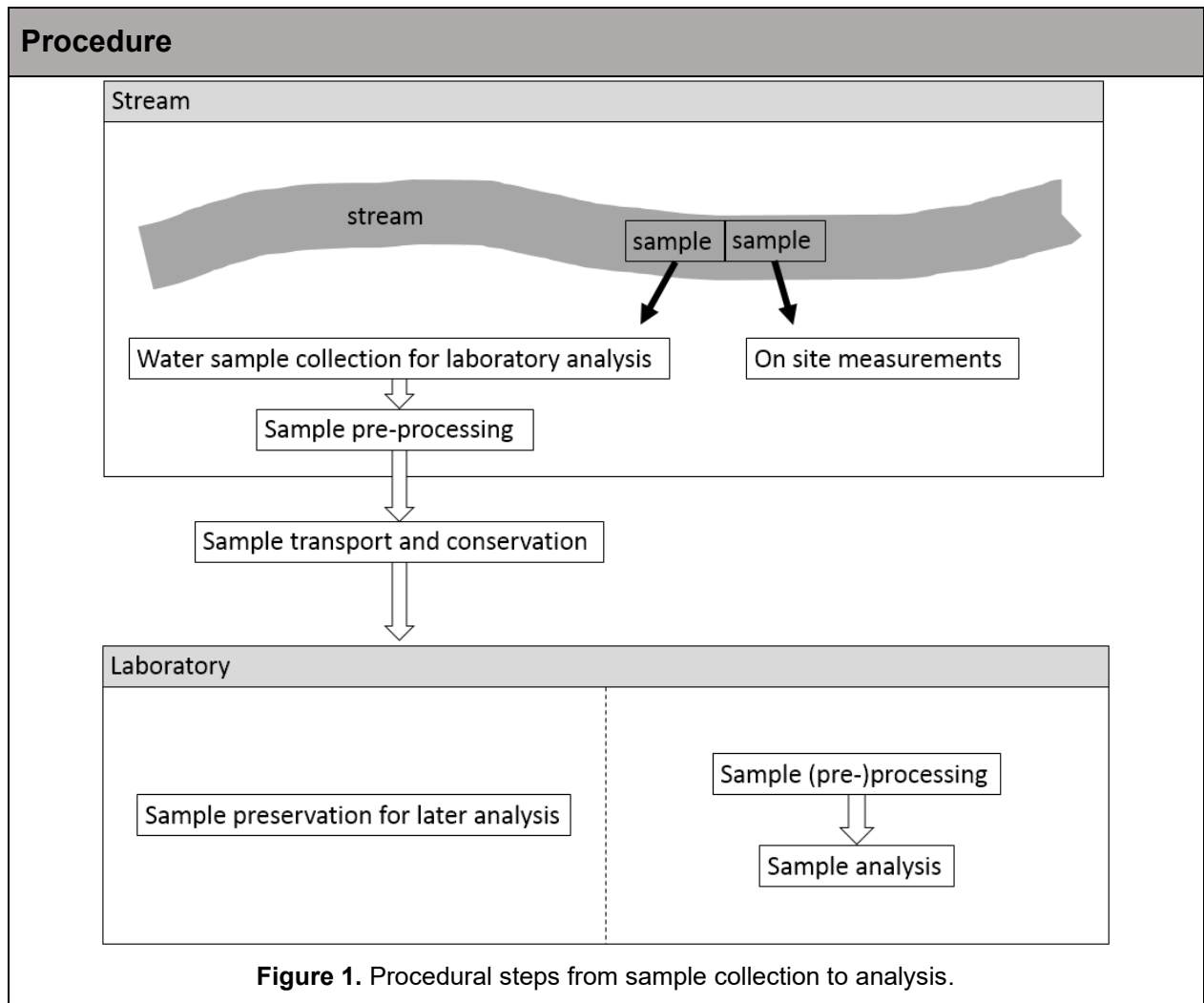
STANDARD OPERATING PROCEDURE

Measurement of Water Quality from Shrimp Farms and Inland Waters

Version	1.0	Date of Issue	05-12-2025										
Purposes	<p>This Standard Operating Procedure (SOP) describes standardized methods for on-site measurement and laboratory-based analysis of water quality variables relevant to aquaculture systems, especially shrimp ponds, canals, and adjacent waterways. It outlines pre-field preparation, field sampling, sample preservation, transport, storage, laboratory analysis, and quality assurance/quality control (QA/QC).</p>												
Scope	<p>This SOP applies to:</p> <ul style="list-style-type: none"> • Shrimp farms (intensive, semi-intensive, rice-shrimp, mangrove-shrimp) • Surrounding natural waters (streams, rivers, canals) • Farmer-led citizen-science measurements • Research monitoring programs requiring reproducible and comparable data <p>It covers two categories of measurements:</p> <ol style="list-style-type: none"> 1. In-situ measurements using multiparameter field probes. 2. Laboratory analyses of collected water samples for nutrient, carbon, and organic matter concentrations. 												
Responsibilities	<p>Before conducting a monitoring campaign, define:</p> <ul style="list-style-type: none"> • Objectives of monitoring (baseline, seasonal variation, pollution hotspot identification, farm-operation monitoring, etc.) • Variables to measure (Table 1) • Monitoring frequency (daily, weekly, per crop cycle stage) • Sampling sites and spatial resolution • Laboratory capabilities and holding-time constraints • Documentation and QA/QC responsibilities <p>Table 1. Water Quality Variables</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="width: 50%;">In-situ (Field Probes)</th> <th style="width: 50%;">Laboratory Analysis</th> </tr> </thead> <tbody> <tr> <td>Temperature</td> <td>Ammonium (NH₄⁺-N)</td> </tr> <tr> <td>Dissolved oxygen (mg/L, % sat.)</td> <td>Nitrite (NO₂⁻-N)</td> </tr> <tr> <td>pH</td> <td>Nitrate (NO₃⁻-N)</td> </tr> <tr> <td>Electrical conductivity</td> <td>Total nitrogen (TN)</td> </tr> </tbody> </table>			In-situ (Field Probes)	Laboratory Analysis	Temperature	Ammonium (NH ₄ ⁺ -N)	Dissolved oxygen (mg/L, % sat.)	Nitrite (NO ₂ ⁻ -N)	pH	Nitrate (NO ₃ ⁻ -N)	Electrical conductivity	Total nitrogen (TN)
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	Turbidity	Ortho-phosphate (PO ₄ ³⁻ -P)
	Chlorophyll-a fluorescence	Total phosphorus (TP)
	Date	Total carbon (TC), TIC, TOC
	Time	Chemical oxygen demand (COD)
	Site metadata	Biochemical oxygen demand (BOD)
Prerequisites	<p>Before conducting a monitoring campaign, define:</p> <ul style="list-style-type: none"> • Objectives of monitoring (baseline, seasonal variation, pollution hotspot identification, farm-operation monitoring, etc.) • Variables to measure (Table 1) • Monitoring frequency (daily, weekly, per crop cycle stage) • Sampling sites and spatial resolution • Laboratory capabilities and holding-time constraints • Documentation and QA/QC responsibilities <p>Responsibilities of each team:</p> <ul style="list-style-type: none"> • Field Team: Site selection, on-site measurements, water sampling, correct labelling, proper preservation. • Laboratory Team: Sample logging, storage, processing, analyses, QC/QA. • Data Manager: Ensures standardized data entry, annotation, and metadata archiving. <p>Required equipment and materials</p> <p>Field Measurement Equipment</p> <ul style="list-style-type: none"> • Multiparameter probe (e.g., YSI 6-series or equivalent) • Calibration solutions: <ul style="list-style-type: none"> ○ pH 4, 7; conductivity 1412 μS/cm; turbidity 0 & 100 NTU; DO barometric calibration • Large bucket (15–20 L), pitcher, rinse water • Waders or boots, long gloves • Field notebook + waterproof pens • Barometer or built-in atmospheric pressure sensor • Cooling boxes with ice packs <p>Sample Collection Materials</p> <ul style="list-style-type: none"> • HDPE bottles (50 mL) • Amber glass bottles (10–50 mL, carbon analyses) • Syringe filters (0.45 μm PES), syringes (50 mL) 	

	<ul style="list-style-type: none"> • Sulfuric acid (H₂SO₄) for COD preservation • Labels, permanent markers • Disposable gloves • Bucket & pitcher for sample homogenization
Related documents	<ol style="list-style-type: none"> 1. APHA, 2022. Standard methods for the examination of water and wastewater. American Public Health Association (APHA): Washington, DC, USA. 2. Ho, L., Thas, O., Van Echelpoel, W., Goethals, P., 2019. A Practical Protocol for the Experimental Design of Comparative Studies on Water Treatment. Water 11, 162. 3. van Loosdrecht, M.C., Nielsen, P.H., Lopez-Vazquez, C.M., Brdjanovic, D., 2016. Experimental Methods in Wastewater Treatment. IWA Publishing



Step 1: Pre-Field Preparations	<ol style="list-style-type: none"> 1. Check and calibrate probes according to manufacturer instructions and frequencies (pH, DO, EC, turbidity, chlorophyll). Calibration logs must be kept. 2. Prepare sampling kits: <ul style="list-style-type: none"> ○ Label all recipients in advance ○ Pre-preserve COD bottles with sulfuric acid (or prepare acid pipettes for field acidification) 3. Charge electronic devices and verify data-logging settings (recommended: 30 measurements at 5-second intervals). 4. Freeze ice packs and prepare cooling boxes. 5. Print field protocols and check access permissions for sampling sites.
Step 2: Field Measurements (In-situ Probes)	<ol style="list-style-type: none"> 1. Site Selection (per visit) Choose a location that is: <ul style="list-style-type: none"> • Well-mixed and representative • Free from sediment disturbance, vegetation, or recent livestock/farmer activity • Safe to access 2. Measurement Steps <ol style="list-style-type: none"> 1. Calibrate DO sensor for local barometric pressure. 2. Rinse gloves and bucket 2–3× with sample water. 3. Collect water in a bucket unless conditions allow safe direct in-stream measurement. 4. Rinse probe sensors with sample water. 5. Immerse probe at least 10 cm depth; allow 2–3 minutes temperature acclimation. 6. Stabilize readings for 2–10 minutes. 7. Log ~30 measurements at 5-second intervals. 8. Record the final values in the field notebook as backup. <p>Important:</p> <ul style="list-style-type: none"> • Avoid aerating the sample. • Do not expose sensors to debris or turbulence. • Keep the bucket shaded to reduce temperature drift
Step 3: Water Sample Collection for	<ol style="list-style-type: none"> 1. General Collection Rules <ul style="list-style-type: none"> • Wear clean disposable gloves; avoid contact between skin and sample.

Laboratory Analyses	<ul style="list-style-type: none"> • Homogenize water in the bucket gently using the pitcher. • Never submerge bottles directly into the bucket. • Rinse bottles 2–3× with sample water (unless pre-preserved). • Fill bottles as required (with or without headspace). • Store immediately in cooling box at 1–4°C and in the dark. <p>2. Filtration</p> <p>For nutrients (NH₄⁺, NO₂⁻, NO₃⁻, PO₄³⁻):</p> <ol style="list-style-type: none"> 1. Fill syringe from pitcher; 2. Rinse syringe (no filter attached) with sample; 3. Attach new 0.45 µm PES filter; 4. Discard first 2 mL filtrate; 5. Rinse sample bottle with filtrate; 6. Fill bottle with filtered sample. <p>3. Preservation Overview</p> <table border="1" data-bbox="448 955 1417 1333"> <thead> <tr> <th>Variable Group</th> <th>Preservation</th> <th>Holding Time</th> </tr> </thead> <tbody> <tr> <td>NH₄⁺, NO₂⁻</td> <td>Cool 1–4°C; analyze ASAP (<24h)</td> <td><24h or freeze (-20°C) 1 month</td> </tr> <tr> <td>NO₃⁻, PO₄³⁻</td> <td>Cool 1–4°C</td> <td><24h or freeze (-20°C)</td> </tr> <tr> <td>TN, TP</td> <td>Cool and dark</td> <td><24h or freeze</td> </tr> <tr> <td>COD</td> <td>Acidify to pH≤2 with H₂SO₄</td> <td>Up to 7 days cooled or freeze</td> </tr> <tr> <td>BOD</td> <td>No preservation; analyze immediately (<6h)</td> <td>If needed: <24h cooled</td> </tr> <tr> <td>TOC/TIC/TC</td> <td>Glass, PTFE-lid; analyze immediately</td> <td>Freeze only if necessary</td> </tr> </tbody> </table>	Variable Group	Preservation	Holding Time	NH ₄ ⁺ , NO ₂ ⁻	Cool 1–4°C; analyze ASAP (<24h)	<24h or freeze (-20°C) 1 month	NO ₃ ⁻ , PO ₄ ³⁻	Cool 1–4°C	<24h or freeze (-20°C)	TN, TP	Cool and dark	<24h or freeze	COD	Acidify to pH≤2 with H ₂ SO ₄	Up to 7 days cooled or freeze	BOD	No preservation; analyze immediately (<6h)	If needed: <24h cooled	TOC/TIC/TC	Glass, PTFE-lid; analyze immediately	Freeze only if necessary
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Step 4: Transport & Storage	<ol style="list-style-type: none"> 1. Transport all samples in cooling boxes (1–4°C). 2. Keep all glass bottles protected from impacts and light. 3. Upon arrival, separate samples for: <ul style="list-style-type: none"> • Immediate analysis (≤6 h) • Next-day analysis (≤24 h) • Long-term storage (freezing -20°C) 																					
Step 5: Laboratory Procedures	<p>1. Sample Processing</p> <ol style="list-style-type: none"> 1. Split collective sample (if taken) into separate bottles for Groups A–C. 2. Filter nutrient samples immediately if not already done in the field. 																					

	<p>3. Allow reagent kits and samples to reach room temperature before analysis.</p> <p>2. Spectrophotometric Analysis (Hach/Lange cuvette tests)</p> <ul style="list-style-type: none"> Follow manufacturer instructions. Include one analysis blank and at least one standard per test run. Use test standards within 30–60% of the test detection range. Prepare dilution series for highly contaminated samples. For colored or turbid samples, run sample-specific blanks and correct results accordingly. <p>3. Sample Analysis (use APHA standard methods)</p>
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Quality management													
QA / QC procedures	<p>1. Probe Calibration</p> <p>Sensors must be calibrated:</p> <table border="1"> <thead> <tr> <th>Sensor</th> <th>Frequency</th> </tr> </thead> <tbody> <tr> <td>Conductivity</td> <td>Weekly</td> </tr> <tr> <td>pH</td> <td>Weekly + 2-day verification</td> </tr> <tr> <td>DO</td> <td>Before each field day if elevation changes >50–100 m</td> </tr> <tr> <td>Turbidity</td> <td>Weekly</td> </tr> <tr> <td>Chlorophyll</td> <td>Weekly</td> </tr> </tbody> </table> <p>A calibration log must include:</p> <ul style="list-style-type: none"> Date/time Standard concentration Sensor drift Notes on anomalies <p>2. QC for Laboratory Data</p> <p>Check data reliability using:</p> <p>Blanks</p> <ul style="list-style-type: none"> Analysis blank: deionized water; detects contamination and baseline drift. Sample-specific blank: detects color/turbidity interference. <p>Standards</p> <p>A result is reliable if standard values fall within $\pm 20\%$ of expected performance (efficiency-corrected).</p> <p>Plausibility Checks</p>	Sensor	Frequency	Conductivity	Weekly	pH	Weekly + 2-day verification	DO	Before each field day if elevation changes >50–100 m	Turbidity	Weekly	Chlorophyll	Weekly
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	<ul style="list-style-type: none"> • $TN \geq \text{sum}(\text{NH}_4^+ + \text{NO}_2^- + \text{NO}_3^-)$ • $TP \geq \text{PO}_4^{3-}$ • $TC \geq \text{TIC}$ and $TC \geq \text{TOC}$ • $\text{COD} \geq \text{BOD}$ <p>Corrections</p> <ul style="list-style-type: none"> • Turbidity sensor correction: add +0.5 NTU or the most negative recorded value to ensure true minimum = 0 NTU. • Sample blank subtraction required when color interference occurs. <p>3. Documentation</p> <p>Field notebooks must record:</p> <ul style="list-style-type: none"> • Date, time, GPS coordinates • Environmental observations • Probe calibration verification • Sampling conditions • Deviations from SOP <p>Laboratory notebooks must record:</p> <ul style="list-style-type: none"> • Holding times • Standard and blank results • Dilutions, spikes, errors • Observed anomalies with cuvette tests
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Version Log

Version	Authors	Changes	Release Date
1.0	Long Ho	Initial SOP for Water Quality Monitoring	05-12-2025

Bibliography

APHA, 2022. Standard methods for the examination of water and wastewater. American Public Health Association (APHA): Washington, DC, USA.

Ho, L., Thas, O., Van Echelpoel, W., Goethals, P., 2019. A Practical Protocol for the Experimental Design of Comparative Studies on Water Treatment. Water 11, 162.

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